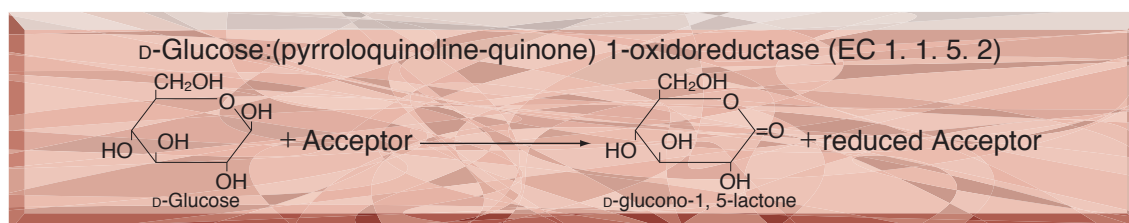


● **TOYOBO ENZYMES** ●  
**(Diagnostic Reagent Grade)**

# GLUCOSE DEHYDROGENASE (PQQ-dependent)

*from Microorganism*



## PREPARATION and SPECIFICATION

Appearance	: Purple amorphous powder, lyophilized
Activity	: Grade III 500U/mg-solid or more
Contaminants	: Glucose dehydrogenase (NAD-dependent) $\leq 1.0 \times 10^{-3}\%$ Hexokinase $\leq 1.0 \times 10^{-3}\%$
Stabilizers	: Ca <sup>++</sup> , BSA

## PROPERTIES

Stability	: Stable at $-20^{\circ}\text{C}$ for at least one year	(Fig.1)
Molecular weight	: approx. 100,000 (by gel filtration)	
Michaelis constant	: 4.8mM (D-Glucose)	
Inhibitors	: Cu <sup>++</sup> , Pb <sup>++</sup> , Ag <sup>+</sup>	
Optimum pH	: 7.0	(Fig.2)
Optimum temperature	: $37^{\circ}\text{C}$	(Fig.3)
pH Stability	: pH 3.5–8.5 ( $25^{\circ}\text{C}$ , 16hr)	(Fig.4)
Thermal stability	: below $50^{\circ}\text{C}$ (pH 7.5, 30min)	(Fig.5)
Substrate specificity	: (Table 1)	
Effect of various chemicals	: (Table 2)	

## APPLICATIONS

This enzyme is useful for enzymatic determination of D-Glucose.

## ASSAY

### Principle:

D-glucose + PMS  $\xrightarrow{\text{PQQ-glucose dehydrogenase}}$  D-glucono-1, 5-lactone + PMS (red)

2PMS (red) + NTB  $\longrightarrow$  2PMS + Diformazan

The appearance of diformazan formed by the reduction of nitrotetrazorium blue (NTB) with phenazine methosulfate (PMS)(red) is measured at 570nm by spectrophotometry.

### Unit definition:

One unit causes the formation of one half micromole of diformazan per minute under the conditions described below.

### Method:

#### Reagents

- A. D-Glucose solution : 1M [1.8g D-Glucose (MW=180.16)/10ml H<sub>2</sub>O] keep this solution at room temperature at least 3 hours before use
- B. PIPES-NaOH buffer, pH 6.5 : 50mM [Weight 1.51g of PIPES (MW=302.36), suspended in 60ml of H<sub>2</sub>O, dissolve with 5N NaOH and add 2.2ml of 10% Triton X-100. After adjusting pH to 6.5±0.05 at 25°C with 5N NaOH, fill up to 100ml with H<sub>2</sub>O]
- C. PMS solution : 3.0mM [9.19mg Phenazine methosulfate (MW=306.34)/10ml H<sub>2</sub>O]
- D. NTB solution : 6.6mM [53.96mg nitrotetrazorium blue (MW=817.65)/10ml H<sub>2</sub>O]
- E. Enzyme diluent : 50mM PIPES-NaOH buffer, pH 6.5 containing 1mM CaCl<sub>2</sub>, 0.1% Triton X-100, 0.1% BSA

#### Procedure

1. Prepare the following reaction mixture in a brownish

bottle and store on ice. (Prepare freshly)

- |        |                           |     |
|--------|---------------------------|-----|
| 0.9ml  | D-Glucose solution        | (A) |
| 25.5ml | PIPES-NaOH buffer, pH 6.5 | (B) |
| 2.0ml  | PMS solution              | (C) |
| 1.0ml  | NTB solution              | (D) |

Concentration in assay mixture	
PIPES-buffer	42 mM
D-Glucose	30 mM
PMS	0.20mM
NTB	0.22mM

2. Pipet 3.0ml of working solution into a test tube (plastic tube) and equilibrate at 37°C for about 5 minutes.
3. Add 0.1ml of enzyme solution\* and mix by gentle inversion.
4. Record the increase of optical density at 570nm against water for 4 to 5 minutes in a spectrophotometer thermostated at 37°C, and calculate the  $\Delta OD$  per minute from the initial linear portion of the curve ( $\Delta OD$  test).

At the same time, measure the blank rate ( $\Delta OD$  blank) by the same method as test except that the enzyme diluent (E) is added instead of the enzyme solution.

- \* Dissolve the enzyme preparation on ice cold enzyme diluent (E) and dilute to 0.1–0.8U/ml with the same buffer, immediately before assay. (The use of plastic tube is recommended because of sticky nature.)

#### Calculation

Activity can be calculated by using the following formula :

$$\text{Volume activity (U/ml)} = \frac{\Delta OD/\text{min} (\Delta OD \text{ test} - \Delta OD \text{ blank}) \times V_t \times df}{20.1 \times 1.0 \times V_s} = \Delta OD \times 1.54 \times df$$

$$\text{Weight activity (U/mg)} = (\text{U/ml}) \times 1/C$$

V<sub>t</sub> : Total volume (3.1ml)

V<sub>s</sub> : Sample volume (0.1ml)

20.1 : Half a millimolar extinction coefficient of diformazan (cm<sup>2</sup>/0.5 micromole)

1.0 : Light path length (cm)

df : Dilution factor

C : Enzyme concentration in dissolution (c mg/ml)

## REFERENCES

- 1) K.Matsushita et al.; *FEMS Microbiology Letters*, 55, 53 (1988).

Table 1. Substrate Specificity of PQQ-Glucose dehydrogenase

Substrate (50mM)	Relative activity(%)	Substrate (50mM)	Relative activity(%)
D-Glucose	100.0	Galactose	16.0
L-Glucose	0.3	D-Lactose	68.9
D-Xylose	15.0	D-Sorbitole	0.2
2-Deoxy-glucose	4.9	D-Mannitol	0.0
L-Sorbose	0.5	Sucrose	0.2
D-Mannose	10.8	Inositol	0.0
D-Fructose	0.3	Maltose	107.0

Table 2. Effect of Various Chemicals on PQQ-Glucose dehydrogenase

[The enzyme dissolved in 50mM PIPES-NaOH buffer, pH 6.5 contg. 1mM CaCl<sub>2</sub>, 0.1% Triton X-100 (5U/ml) was incubated with each chemical at 25°C for 1hr.]

Chemical	Concn.(mM)	Residual activity(%)	Chemical	Concn.(mM)	Residual activity(%)
None	—	100	MIA	2.0	87
Metal salt	2.0		NEM	2.0	100
MgSO <sub>4</sub>		108	IAA	2.0	98
CaCl <sub>2</sub>		108	Hydroxylamine	2.0	19
Ba(OAc) <sub>2</sub>		105	EDTA	5.0	79
FeCl <sub>3</sub>		79	O-Phenanthroline	2.0	7
CoCl <sub>2</sub>		42	$\alpha, \alpha'$ -Dipyridyl	1.0	103
MnCl <sub>2</sub>		105	Borate	5.0	110
ZnCl <sub>2</sub>		45	NAF	2.0	111
Cd(OAc) <sub>2</sub>		107	NaN <sub>3</sub>	2.0	115
NiCl <sub>2</sub>		101	Triton X-100	0.10%	101
CuSO <sub>4</sub>		0	Brij 35	0.10%	22
Pb(OAc) <sub>2</sub>		0	Tween 20	0.10%	104
AgNO <sub>3</sub>		0	Span 20	0.10%	60
HgCl <sub>2</sub>		77	Na-Cholate	0.10%	67
2-Mercaptoethanol	2.0	99	SDS	0.05%	33
PCMB	1.0	97	DAC	0.05%	113

Ac, CH<sub>3</sub>CO; PCMB, p-Chloromercuribenzoate; MIA, Monoiodoacetate; NEM, N-Ethylmaleimide; IAA, Iodoacetamide; EDTA, Ethylenediaminetetraacetate; SDS, Sodium dodecyl sulfate; DAC, Dimethylbenzylalkylammonium chloride.

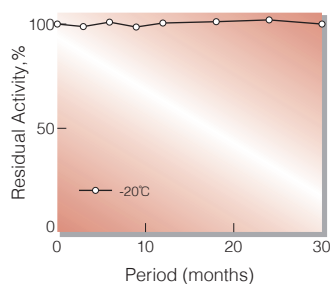


Fig. 1. Stability (Powder form)  
[kept under dry conditions]

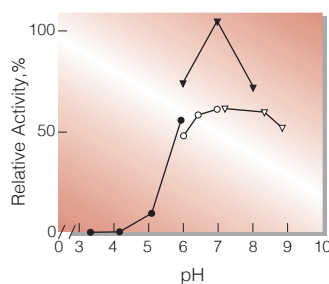


Fig. 2. pH-Activity

[25°C, in 50mM buffer solution;  
●—●, acetate; ▼—▼, phosphate;  
○—○, PIPES; ▽—▽, Tris-HCl.]

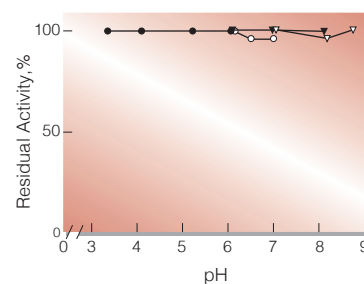


Fig. 4. pH-Stability

[25°C, 16 hr-treatment with 50mM buffer solution  
contg. 1mM CaCl<sub>2</sub>; ●—●, acetate;  
▼—▼, phosphate; ○—○, PIPES; ▽—▽, Tris-HCl.]

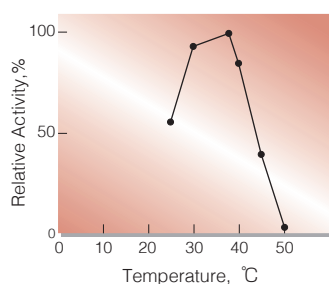


Fig. 3. Temperature Activity

[in 42mM PIPES-NaOH buffer, pH 6.5]

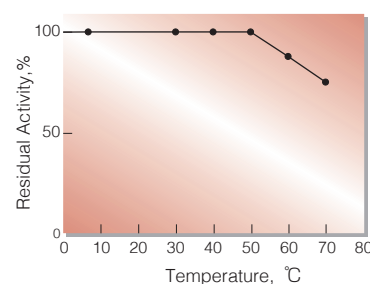


Fig. 5. Thermal stability

[30min.-treatment with 50mM PIPES-NaOH buffer,  
pH 6.5 contg. 1mM CaCl<sub>2</sub> enzyme  
concentration: 5.0 U/ml]

## 活性測定法 (Japanese)

### 1.原理

D-Glucose + PMS  $\xrightarrow{\text{PQQ-glucose dehydrogenase}}$   
 D-glucono-1, 5-lactone + PMS(red)  
 2PMS(red) + NTB  $\longrightarrow$  2PMS + Diformazan  
 PMS(phenazine methosulfate)を介してNTB  
 (nitrotetrazorium blue)を還元し,生成したdiformazan  
 の570nmにおける吸光度を測定する。

### 2.定義

下記条件下で1分間あたり1/2マイクロモルの  
 diformazanを生成する酵素量を1単位(U)とする。

### 3.試薬

- 1M D-グルコース溶液(1.8gのD-グルコースを蒸留水に溶解し,10mlとし,室温で3時間経過したものを使用する)
- 50mM PIPES-NaOH緩衝液,pH6.5(1.51gのPIPESを60mlの蒸留水に懸濁し,5N NaOHで溶解後,10%のTritonX-100溶液2.2mlを加える。5N NaOHを使って25°CでpHを6.5±0.05に調整後,蒸留水で100mlとする)
- 3mM PMS溶液(9.19mgのPMSを10mlの蒸留水に溶解する)
- 6.6mM NTB溶液(53.96mgのNTBを10mlの蒸留水に溶解する)

酵素溶液：酵素標品を予め氷冷した1mM CaCl<sub>2</sub>と0.1% TritonX-100と0.1% BSAを含む50mM PIPES-NaOH緩衝液,pH6.5で0.1~0.8U/mlに希釈する。

### 4.手順

- 褐色瓶中に下記反応混液を調製し,氷冷保存する。  
 (用時調製)
 

0.9ml	D-グルコース溶液	(A)
25.5ml	PIPES-NaOH緩衝液	(B)
2.0ml	PMS溶液	(C)
1.0ml	NTB溶液	(D)
- 反応混液3.0mlをプラスチック製のキュベット(d=1.0cm)に分注し37°Cで約5分間予備加温する。
- 酵素溶液を0.1mlを添加し,緩やかに混和後,水を対照に37°Cに制御された分光光度計で570nmの吸光度変化を4~5分間記録し,その直線部分から1分間あたりの吸光度変化を求める(ΔODtest)。
- 盲検は反応混液①に酵素液の代わりに酵素希釈液(1mM CaCl<sub>2</sub>と0.1% TritonX-100と0.1% BSAを含む50mM PIPES-NaOH緩衝液, pH6.5)0.1mlを加え,上記同様に操作を行い,1分間あたりの吸光度変化を求める(ΔODblank)。

### 5.計算式

$$U/ml = \frac{\Delta OD/min (\Delta OD \text{ test} - \Delta OD \text{ blank}) \times 3.1 \times df}{20.1 \times 1.0 \times 0.1}$$

$$= \Delta OD/min \times 1.54 \times df$$

$$U/mg = U/ml \times 1/C$$

20.1 : diformazanの1/2ミリモル分子吸光係数  
 (cm<sup>2</sup>/0.5 micromole)

1.0 : 光路長(cm)

C : 溶解時の酵素濃度(C mg/ml)