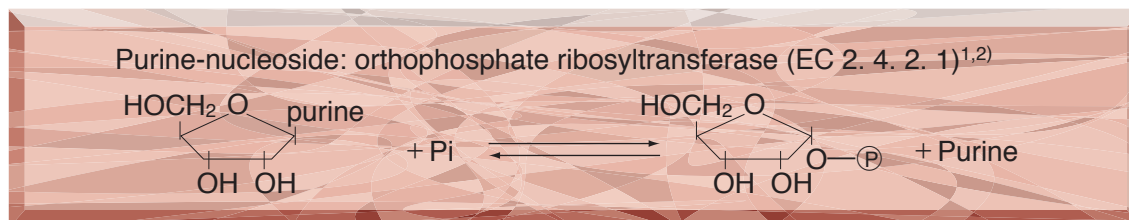


● **TOYOBO ENZYMES** ●  
**(Diagnostic Reagent Grade)**

# PURINE-NUCLEOSIDE PHOSPHORYLASE

*from Microorganism*



## PREPARATION and SPECIFICATION

Appearance	: White amorphous powder, lyophilized
Activity	: Grade III 15U/mg-solid or more
Contaminants	: Catalase ≤20%
	: 5'-Nucleosidase ≤1.0×10 <sup>-3</sup> %
	: Adenosine deaminase ≤1.0×10 <sup>-3</sup> %
	: ATPase ≤1.0×10 <sup>-2</sup> %
Stabilizers	: K-Gluconate, mannitol, EDTA

## PROPERTIES

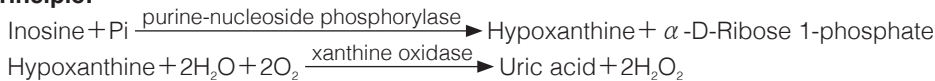
Stability	: Stable at -20°C for at least one year	(Fig.1)
Molecular weight	: approx. 120,000	
Isoelectric point	: 4.1±0.1	
Michaelis constant	: 6.4×10 <sup>-5</sup> M (Inosine), 3.2×10 <sup>-4</sup> M (Pi)	
Inhibitors	: p-Chloromercuribenzoate, SDS, Hg <sup>++</sup> , Ag <sup>+</sup>	
Optimum pH	: 7.5-8.0	(Fig.3)
Optimum temperature	: 65°C	(Fig.4)
pH Stability	: pH 6.0-9.0 (30°C, 16hr)	(Fig.5)
Thermal stability	: below 60°C (pH 7.7, 30min)	(Fig.6)
Substrate specificity	: (Table 1)	
Effect of various chemicals	: (Table 2)	

## APPLICATIONS<sup>3~5)</sup>

This enzyme is useful for enzymatic determination of inorganic phosphorus, 5'-nucleotidase and adenosine deaminase when coupled with xanthine oxidase (XTO-212) and uricase (UAO-201, UAO-211)

## ASSAY

### Principle:



The appearance of uric acid is measured at 293nm by spectrophotometry.

### Unit definition:

One unit causes the formation of one micromole of uric acid per minute under the conditions describe below.

### Method:

#### Reagents

- A. K-Phosphate buffer, pH 7.7 : 50mM  
 B. Inosine solution : 32mM [Dissolve 85.8mg of inosine (MW=268.23) in 10ml of H<sub>2</sub>O with heating]  
 (Stable for at least two weeks if stored at 4°C)  
 C. Xanthine oxidase solution : ca.6.6U/ml [Dissolve xanthine oxidase (XTO-212) to ca.6.6U/ml with ice-cold  
 buffer A] (Should be prepared fresh)  
 D. Enzyme diluent : buffer A

#### Procedure

- Prepare the following reaction mixture in a cuvette (d=1.0cm) and equilibrate at 37°C for about 5 minutes.
 

Concentration in assay mixture	
K-Phosphate buffer	ca.47 mM
Inosine	2.1 mM
Xanthine oxidase	ca. 0.2U/ml

2.7ml K-Phosphate buffer, pH 7.7 (A)  
 0.2ml Substrate solution (B)  
 0.1ml Xanthine oxidase solution (C)
- Add 0.05ml of the enzyme solution\* and mix by gentle inversion.
- Record the decrease in optical density at 293nm against water for 3 to 4 minutes in a spectrophotometer thermostated at 37°C, and calculate the  $\Delta\text{OD}$  per minute from the initial linear portion of the curve ( $\Delta\text{OD}$  test).  
 At the same time, measure the blank rate ( $\Delta\text{OD}$  blank) by using the same method as the test except that the enzyme diluent is added instead of the enzyme solution.

\* Dissolve the enzyme preparation in ice-cold enzyme diluent (D), and dilute to 0.1–1.5U/ml with the same buffer and store on ice.

#### Calculation

Activity can be calculated by using the following formula :

$$\text{Volume activity (U/ml)} = \frac{\Delta\text{OD}/\text{min} (\Delta\text{OD test} - \Delta\text{OD blank}) \times V_t \times \text{df}}{12.5 \times 1.0 \times V_s} = \Delta\text{OD}/\text{min} \times 4.88 \times \text{df}$$

$$\text{Weight activity (U/mg)} = (\text{U/ml}) \times 1/\text{C}$$

V<sub>t</sub> : Total volume (3.05ml)

V<sub>s</sub> : Sample volume (0.05ml)

12.5 : Millimolar extinction coefficient of uric acid under the assay condition (cm<sup>2</sup>/micromole)

1.0 : Light path length (cm)

df : Dilution factor

C : Enzyme concentration in dissolution (c mg/ml)

## REFERENCES

- 1) R.E.Parks, Jr. and R.P.Agarwal; *The Enzymes, Vol.7*, p483 (3rd ed.)(1972)
- 2) P.A.Hoffe, R.May and B.C.Robertson; *Methods in Enzymology, Vol.11*, p70 (1972)
- 3) Y.Machida and T. Nakanishi; *Agric.Biol.Chem.,45*, 1801 (1981)
- 4) M.Sugiura, K.Kato, T.Adachi, Y.Ito, K.Hirano and S.Sawaki; *Chem.Pharm.Bull.,29*, 1451 (1981)
- 5) P.Fossati; *Analytical Biochemistry.,149*, 62 (1985)

**Table 1. Substrate Specificity of Purine-nucleoside phosphorylase<sup>3)</sup>**

[Inosine: Purine-nucleoside phosphorylase - Xanthine oxidase system, pH 7.7  
Guanosine, Adenosine, ATP, Thymidine: UV-system, pH 7.4]

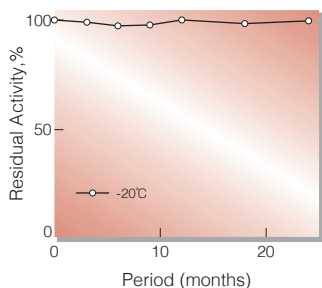
Substrate(0.2mM)	Relative activity(%)	Substrate(0.2mM)	Relative activity(%)
Inosine	100	ATP	0
Guanosine	41	Thymidine	0
Adenosine	0		

**Table 2. Effect of Various Chemicals on Purine-nucleoside phosphorylase**

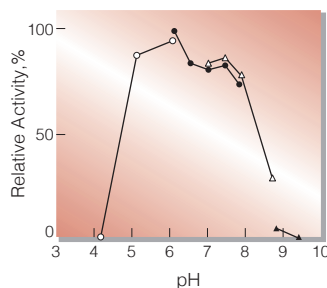
[The enzyme dissolved in 50mM PIPES buffer, pH 7.0 (10U/ml) was incubated with each chemical at 30°C for 1hr.]

Chemical	Concn.(mM)	Residual activity(%)	Chemical	Concn.(mM)	Residual activity(%)
None	—	100	N-ethylmaleimide	2.0	91.9
Metal salt	2.0		NaF	2.0	90.9
MgCl <sub>2</sub>		90.1	NaN <sub>3</sub>	20	95.7
CaCl <sub>2</sub>		96.7	EDTA	5.0	96.8
Ba(OAc) <sub>2</sub>		93.4	o-Phenanthroline	2.0	98.3
FeCl <sub>3</sub>		73.9	Borate	50	9.0
MnCl <sub>2</sub>		95.0	Iodoacetamide	2.0	98.7
ZnCl <sub>2</sub>		77.6	Triton X-100	0.10%	138.4
NiCl <sub>2</sub>		94.1	Na-cholate	0.10%	124.4
CuSO <sub>4</sub>		9.8	SDS	0.10%	0.1
Pb(OAc) <sub>2</sub>		9.1	Span 20	0.10%	128.9
AgNO <sub>3</sub>		0.5			
HgCl <sub>2</sub>		0.1			

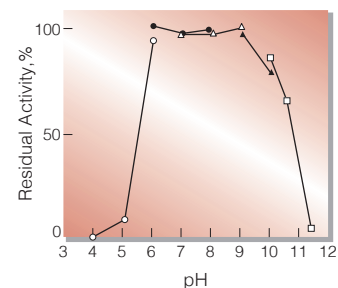
EDTA, Ethylenediaminetetraacetate; SDS, Sodium dodecyl sulfate.



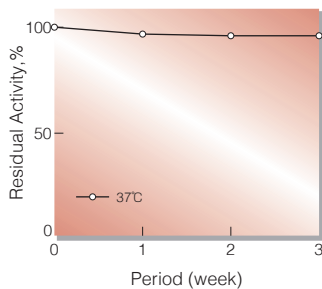
**Fig.1. Stability (Powder form)**  
[kept under dry conditions]



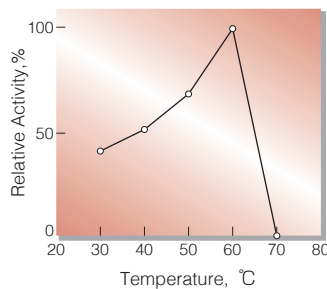
**Fig.3. pH-Activity**  
[ in 50mM buffer solution: pH 4-6, acetate; pH 6-8, K-phosphate; pH 7-9, Tris-HCl; pH 9-10, Borate ]



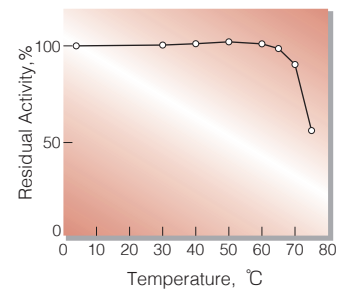
**Fig.5. pH-Stability**  
[ 30°C, 16hr-treatment with 50mM buffer solution: pH 4-6, Acetate; pH 6-8, K-phosphate; pH 7-9, Tris-HCl; pH 9-10, Borate; pH10-12, Glycine-NaOH. Enzyme concentration: 10U/ml ]



**Fig.2. Stability (Powder form)**  
[kept under dry conditions]



**Fig.4. Temperature activity**  
[ in 50mM K-phosphate buffer, pH 7.7 ]



**Fig.6. Thermal stability**  
[ 30min-treatment with 50mM K-phosphate buffer, pH 7.7. Enzyme concentration: 1U/ml ]

## 活性測定法 (Japanese)

### 1.原理

Inosine + Pi  $\xrightarrow{\text{purine-nucleoside phosphorylase}}$  Hypoxanthine +  
 $\alpha$ -D-Ribose 1-phosphate  
 Hypoxanthine + 2H<sub>2</sub>O + 2O<sub>2</sub>  $\xrightarrow{\text{xanthine oxidase}}$  Uric acid +  
 2H<sub>2</sub>O<sub>2</sub>  
 尿酸の生成量を293nmにおける吸光度の変化で測定する。

### 2.定義

下記条件下で1分間に1マイクロモルの尿酸を生成する酵素量を1単位(U)とする。

### 3.試薬

- A. 50mM K-リン酸緩衝液, pH 7.7  
 B. 32mM イノシン水溶液 [85.8mgのイノシン(MW=268.23)を10mlの蒸留水に加温溶解する] (4°C保存で2週間は使用可能)  
 C. キサンチンオキシダーゼ溶液 [Roche製硫酸安懸濁結晶酵素(約20U/ml)を氷冷緩衝液Aで約6U/mlに希釈する] (用時調製)  
 酵素溶液: 酵素標品を予め氷冷した50mM K-リン酸緩衝液, pH7.7で溶解し, 同緩衝液で0.1~1.5U/mlに希釈して氷冷保存する。

### 4.手順

- ① 下記反応混液をキュベット(d=1.0cm)に調製し, 37°Cで約5分間予備加温する。  
 2.7 ml K-リン酸緩衝液 (A)  
 0.20ml 基質溶液 (B)  
 0.10ml キサンチンオキシダーゼ溶液 (C)  
 ② 酵素溶液0.05mlを添加し, ゆるやかに混和後, 水を対照に37°Cに制御された分光光度計で293nmの吸光度変化を3~4分間記録し, その初期直線部分から1分間当たりの吸光度変化を求める( $\Delta$ ODtest)。  
 ③ 盲検は反応混液①に酵素溶液の代わりに酵素希釈液(50mM K-リン酸緩衝液, pH7.7)を0.05mlを加え, 上記同様に操作を行って, 1分間当たりの吸光度変化を求める( $\Delta$ ODblank)。

### 5.計算式

$$U/ml = \frac{\Delta OD/min (\Delta OD_{\text{test}} - \Delta OD_{\text{blank}}) \times 3.05(ml) \times \text{希釈倍率}}{12.5 \times 1.0 \times 0.05(ml)}$$

$$= \Delta OD/min \times 4.88 \times \text{希釈倍率}$$

$$U/mg = U/ml \times 1/C$$

12.5 : 尿酸のミリモル分子吸光係数 (cm<sup>2</sup>/micromole)  
 1.0 : 光路長(cm)  
 C : 溶解時の酵素濃度(c mg/ml)